COMPUTER ANALYSIS OF NON-ISOTHERMAL TG DATA FOR MECHANISM AND ACTIVATION ENERGY. PART I

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ABSTRACT

A computer program (CP) is presented, based on a non-computer procedure reported in the literature for the determination of mechanism and corresponding activation energy utilizing non-isothermal TG data. This CP is then applied to six sets of TG data. The agreement between estimated and reported results is excellent. Various aspects and limitations of the CP are discussed.

INTRODUCTION

The authors recently presented a computer program whereby one of twelve theoretically possible solid-state decomposition mechanisms could be distinguished utilizing non-isothermal (NI) or isothermal TG data [1]. Zsako [2] also presented a method for the kinetic analysis of NI TG data for mechanism (and for activation energy, E). However, this author did not employ a computer procedure. Instead, Tables were used which consisted of so-called $-\log p(x)$ values corresponding to different temperatures and E-values. Further, analysis of NI TG data was confined to only three possible mechanisms, i.e., $n = 0, 1, 2$, assuming an *n*-order type reaction. In this paper, the authors intend to present a computer program based on the procedure outlined by Zsako and, by applying this program to NI TG data, to distinguish one of ten theoretically possible solid-state decomposition mechanisms and to determine the corresponding value of E.

THEORY

Prior to presenting and discussing the aforementioned computer program, some theoretical background would appear to be apropos.

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It has been previously indicated [3] that

$$
g(\alpha) = ZEp(x)/Rq \tag{1}
$$

where $g(\alpha) = \int_0^{\alpha} d\alpha / f(\alpha)$; Z = pre-exponential factor; R = gas constant; q = constant heating rate; and, $p(x) = e^{-x}/x - \int_x^{\infty} e^{-u} \frac{du}{u}$, where $x = E/RT$. When the common logarithm of eqn. (1) is taken, the result is

$$
B = \log(ZE/Rq) = \log g(\alpha) - \log p(x)
$$
 (2)

where $B = constant$. Further, using a truncated Schlömilch expansion [4]

$$
p(x) = e^{-x}(1/x)[1/(x+2)] \tag{3}
$$

Doyle [5] has indicated that eqn. (3) yields an excellent approximation for the true value of $p(x)$ with values of x between 10 and 50. (We have found that eqn. (3) is still a good approximation up to values of $x = 60$). Since values of x are approximately equal to values of E (kcal mol⁻¹), E-values employed in the computer program were in the range $20-60$ kcal mol⁻¹ (cf. lines 35, 55, 75 and 145 of the BASIC computer program (BCPA) in Appendix).

THE COMPUTER PROGRAM

Before making a run, the value of N (number of alpha- $T(K)$ data pairs) in line 15 should be adjusted, if necessary. Also, the following ten theoretical mechanisms were examined to ascertain which one of them conformed best to the NI TG data: A4, A3, and A2 (random nucleation, Avrami equations); R2 and R3 (phase boundary reaction, cylindrical symmetry and spherical symmetry, respectively); F1 (random nucleation, one nucleus per particle); D1, D2, D3, and D4 (corresponding to 1-dimensional diffusion, 2-dimensional diffusion-cylindrical symmetry, 3-dimensional diffusion-Jander spherical symmetry, and 3-dimensional diffusion-Ginstling/Brounshtein spherical symmetry). The mechanisms are further defined in lines 250 and 255 of the BCPA. In order to reduce to reasonable values the host of computations involved and the corresponding relatively long computer running time (to ca., 5-15 min), only ten arbitrarily chosen mechanisms were tested and values of E were incremented by only 1 kcal mol⁻¹.

Values of E ranging from 20–60 kcal mol⁻¹ were assigned to each of the preceding mechanisms. Further, values of B (see eqn. (2)) were calculated for each pair of alpha- $T(K)$ TG data at any particular E-value. An average value, \overline{B} , was then determined (line 90 of BCPA) and finally the standard deviation of B for each mechanism, denoted as Delta, for any particular E-value. Delta is defined as usual

$$
\text{Delta} = \left[\left(B_i - \overline{B} \right)^2 / N \right]^{1/2} \tag{4}
$$
\n(see lines 15, 95 of BCPA).

$$
(4)
$$

In the Appendix, a run is depicted which uses data from line 240. From this run, it can be seen that at $E = 20$ kcal mol⁻¹, the probable mechanism, based on the lowest value of Delta, was F1. When the E -value was increased to 21, the probable mechanism changed to D1 since its Delta now possessed the lowest value. These lowest values of Delta, and the corresponding values of E and the mechanism description (lines 250, 255) were stored for final computer analysis (lines 95, 135).

The computer run could be terminated in one of two ways. Thus, if the value of E attained 60 kcal mol⁻¹, all the lowest Delta-values for each E -value from 20–60 were listed along with their corresponding probable mechanisms. Then the lowest of these Delta-values was selected along with the corresponding E -value and mechanism and these final results were displayed (e.g., see last line of run in Appendix). However, in this type of termination, the printout will also indicate that the computer analysis was incomplete since the run was arbitrarily halted at $E = 60$ (line 145). Under these conditions, the results are uncertain and bear further investigation by other procedures.

The second manner in which the run may be terminated is as follows. From the run in the Appendix, it can be observed that, for the order in which the mechanisms are listed, as the E -value increases, the Delta-values which are physically above the Delta of the probable mechanism (DPM) increase, whereas the Delta-values which are physically below the DPM decrease. The value of the DPM may either increase or decrease with increasing E-value. Further, the magnitude of the Delta-values physically above the DPM increase as the physical position away from the DPM increases. The same holds for the magnitude of the Delta-value physically below the DPM. From the preceding, when the DPM of the last mechanism, D3, begins to increase, all the previously listed mechanisms have been tested, and the run simply ends (lines 140, 155-165, 170).

RESULTS AND DISCUSSION

The TG data presented in lines 185-240 of the BCPA were analyzed by the previously-described computer method. In the following are given in order, the line number containing the analyzed data, the computer-calculated and the reported E-values (kcal mol⁻¹), the computer-determined and the reported mechanism, whether or not the analysis was complete (Y/N) , and finally, the reference from which the data was obtained: 185, 29, 28.9, F1, F1, Y, [2]; 195, 26, 28.3, R3, F1, Y, [2]; 205, 22, 22, R3, n = 0.7, Y, [6]; 215-220, 30, 30, D2, D2, Y, [7]; 230, 28, 28, R2, R2, N, [6]; 240, 30, 30, D3, D3, Y, [8]. From the preceding, it can be seen that the agreement between calculated and reported results is excellent, except for the data used in line 195. Thus, Zsako tested only 3 n-order type mechanisms and determined that the data best fitted $n = 1$ (F1). However, the computer procedure yielded an R3-mechanism and a corresponding $E = 26$. These results were checked using another computer procedure previously reported [9]. This procedure afforded values of $n = 0.64$ (equivalent to R3) and $E = 25$ kcal $mol⁻¹$. Also, all analyses were complete except for the analysis of data in line 230.

Some limitations of the program are: E-values estimated are at best accurate to ± 1 kcal mol⁻¹; only 10 mechanisms were tested; and, an incomplete analysis can result. In order to lessen the running time, E-values were incremented by 1 and only 10 mechanisms were tested. If time is not an important consideration, value of E may be incremented by less than 1 kcal $mol⁻¹$ and the number of mechanisms to be tested can be increased. However, in the latter case, in order to maintain the manner in which the DPM changes as previously described, additional mechanisms should be placed in a certain order [10]. Thus, P3 $[A_0(1/2)]$ should be placed between A3 and A2, and A1.5 $[-\log(1 - A)\sqrt{2/3}]$ between A2 and R2, etc. Obviously, the run-time is also dependent on the value of N.

Whether or not an incomplete computer analysis results depends, in part, on the true E-value and also on the initial physical location of the DPM. Thus, from the order of the mechanisms listed in the run in the Appendix, it can be seen that if the initial DPM is at A4, then a large increase in E might be anticipated in order to reach D3. This increase could reach $E = 60$ and thereby cause an incomplete analysis to result. However, when the initial DPM starts at F1 (as in the run in the Appendix), there is less likelihood that $E = 60$ would be attained.

APPENDIX

A BASIC computer program for analysis of non-isothermal TG data for mechanism and activation energy.

```
10 TEXT : HOME 
15 N = ii: REM NO. DATA PRS. 
20 \text{ } Q = 1: FF = 1: EI = 20000
25 DIM ST$ (50), SS$ (50), DL (50), W(N), T(N), B(N), TS$ (50), LD (50), LS (50), AE (50)
30 R = 1.987: F = 2.303: DL(0) = 100: LD(0) = 100:LS(0) = 100:EE = 035 DEF FN P(X) = (X - LOG ((1 / X) * (1 / (X + 2)11) / F: REM SCHLOMI
    LCH APPROXMN. 
37:40 FOR J = 1 TO N: READ W(J), T(J): NEXT J
45 FOR 3 = I TO 10: READ ST$(J):SS$(J) = ST$(JI: NEXT J 
50 :<br>55 E = 0: REM
                 EA=20-60 KCAL/MOL
57 : 
60 FOR CC = 1 TO 10<br>65 ON CC 60SUB 270,280,300,320,330,340,350,360,370,38
67 : 
70 FOR K = 1 TO N
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```
75 EA = (EI + 1000 * (FF * E - 1)): X = EA / (R * T(K))BO B(K) = FN M(W(K)) + FN P(X)B5 NEXT K
B7:90 FOR L = 1 TO N: BA = BA + B(L): NEXT L: BA = BA / N
95
    FOR M = 1 TO N: DI = DI + (B(M) - BA) ^ 2: NEXT M: DL(CC) = SOR (DI / N
     ) + DL(CC)100 \text{ D1} = 0:BA = 0105 NEXT CC
110 EE = EE + 1:LS(EE) = DL(CC - 1)
112:FOR 1 = 1 TO CC - 1
115
    PRINT "FOR "ST$(1);", DELTA= "; INT (DL(I) * 1E4 + .5) / 1E4<br>IF DL(I) < = DL(I - 1) THEN 135
120
125
130 DL(I) = DL(I - 1):ST$(I) = ST$(I - 1)
    NEXT I: PRINT : PRINT "PROB.MECHNSM.= ";ST$(1 - 1)" FOR EA= ";EA / 10
135
     00;" K/M":LD(0) = DL(I - 1):TS$(0) = ST$(I - 1):AE(0) = EA: FOR 3 = 1TO 12: DL (J) = 0: NEXT J: PRINT
     IF LS(EE) > LS(EE - 1) THEN 155
140.145 Q = 0 + 1: IF 0 = 42 THEN FOR 3 = 1 TO 41: PRINT INT (LD(J) * 1E4 +
     .5) / 1E4; "---"; TS$ (J); NEXT J; PRINT : PRINT "ANALYSIS INCOMPLETE!!;
      RESULTS OBTAINED ARE: ": GOTO 155
     FOR K = 1 TO 10:ST*(K) = SS*(K); NEXT K: GOTO 55
150
155 FOR J = 1 TO EE: IF LD(J) \langle = LD(3 - 1) THEN 165
160 \text{ LD}(J) = \text{LD}(J - 1) : TS* (J) = TS* (J - 1) : AE(J) = AE(J - 1)155 NEXT J
167:170 PRINT: PRINT "EA="AE(J - 1) / 1000" KCAL/MOL FOR MECHNSM. ":TS$(J -
     1);", DELTA= ";LD(J - 1)
175:180<sub>1</sub>END
182:185 REM DATA.03398,433.2,.10194,443.2,.19903,453.2,.36893,463.2,.58738,47
     3.2,.86893,483.2,.91748,493.2: REM CMPLX.1, 2SAKO DATA-->EA=29K/M, F1
     -MECH. (7 PRS.)
190 :
    REM DATA .011236,413.2,.022472,423.2,.039326,433.2,.078652,443.2,.191
195.
     01,453.2,.34270,463.2,.58427,473.2,.87640,483.2,.97191,493.2: REM C
     MPLX. II, ZSAKO DATA-->EA=26K/M; R3-MECHNSM. (9-FRS.)
200:
    REM DATA.208,408.2,.3,413.2,.403,418.2,.528,423.2,.667,428.2,.806,433<br>2,.917,438.2: REM REICH, NAHCO3, TA,24,9(1978): DATA-->EA=22 FOR R3.
205
     -MECHNSM. (7-PRS.)
210:215 REM DATA. 04438, 602, .0577, 614, .0E392, 632, .1194, 650, .16635, 668, .22714,
     686, .2765, 698, .3336, 710, .39897, 722, .4348, 728, .4728, 734, .5128, 740, .554
     8.746. 5988.752: REM DATA CONT'D. CN LN #220
217:0414.6444.758..6916.764..73986.770..78884.776..8379.782..8861.78
220
    REM
     8, 9321, 794, 9735, 800: REM HEIDE DATA-->30K/M FOR D2-MECHNSM. (22-PR
     5.3225:REM DATA.1319, 405, .20195, 410, .30261, 415, .44105, 420, .61869, 425, .81878,
230.
                             THEOR.DATA, REICH, EA-->28.0 K/M, R2-MECHNSM.
     430..97883.435: REM
235:1.07197,620,10695,640,15439,660,121661,680,129533,700,139099,7240 DATA
     20, 50186, 740, 62308, 760, 74587, 780, 85746, 800, 94279, 820: REM REIC
     H, TA, 34, 287(1979) (D3-MECH); DATA-->30K/M FOR D3-MECHNSM. (11 PRS.)
245:"A4: (-LN(1-A))^(1/4)","A3: (-LN(1-A))^(1/3)","A2: (-LN(1-A))^
250
     DATA
     (1/2) ", "R2: 1 - (1 - A) \cap (1/2) ", "R3: 1 - (1 - A) \cap (1/3) ", "F1: -LN(1 - A)"
             "D1: A^2", "D2: A+(1-A)LN(1-A)", "D4: 1-(2A/3)-(1-A)^(2/3)", "D3
255
     DATA
     (1-(1-A)^(1/3))^2260 :
                      LOG (( - LOG (1 - X)) ^ (1 / 4)) / F: RETURN
270
     DEF
          FN W(X) =FN W(X) = LOG (( - LOG (1 - X)) ^ (1 / 3)) / F: RETURN
280
     DEF
300 DEF FN W(X) = LOG (( - LOG (1 - X)) ^ (1 / 2)) / F: RETURN
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170

 320 def FN W(X) = LOG (1 – (1 – X) \sim (1 / 2)) / F: RETURN 330 DEF FN W(X) = LOG (1 - (1 - X) \sim (1 / 3)) / F: RETURN LOG ($-$ LOG (1 - X)) / F: RETURN 340 DEF. FN W(X) = LOG $(X \cap 2)$ / F: RETURN 350 DEF FN $W(X) =$ 560. DEF. FN $W(X) =$ LOG (X + (1 − X) * LOG (1 − X)) / F; RETURN
LOG (1 − (2 * X / 3) – (1 − X) ^ (2 / 3)) / F; RETURN 370 DEF FN W(X) = 380 DEF FN W(X) = LOG ((1 - (i - X) " (I / 3)) "" 2) / F: RETURN 3RUN. $|F$ OR A4: (-LN(1-A)) $^{\circ}$ (1/4), DELTA= .4915 $^{\circ}$ OR A3: (-LN(1-A)) $^{\circ}$ (1/3), DELTA= .4506 FOR A2: (-LN(I-A))~(I/2), DELTA= .3689 FOR R2: $1-(1-A) \cap (1/2)$, DELTA= .1962 FOR RS: 1-(1-A)~(1/3), DELTA= .1733
FOR F1: -LN(1-A), DELTA= .1259 FOR DI: A'2, DELTA= .1285 FOR D2: A+fI-A)LN(I-A), DELTA= .1819 FOR D4: !-(2A/3}-(I-A)"(2/3)~ DELTA= .2084 FOR D3: ~I--(I-A)"(I/3)>"2~ DELTA= °2675 PROB.MECHNSM. = F1: - LN(1-A) FOR EA= 20 K/M FOR A4: $(-LM(1-A)) \cap (1/4)$, DELTA= .5188 FOR A3: (-LN(1-A))-(1/3), DELTA= .4779 FOR A2: $(-LN(1-A))$ $(1/2)$, DELTA= .3962 FOR R2: I-(I-A) ~(i/2), DELTA= .2235 $FOR R3: 1-(1-A)^\sim(1/3)$, DELTA= .2006 FOR FI: -LN(I-A)~ DELTA= .1528 FOR Di: An2, DELTA= .107 FOR D2: A+(I-A)L~(I-A)~ DELTA= .1556 FOR D#: I-(2A/3)-(I-A)'-(2/3), DELTA= .1814 SCR D3: $(1-(1-A)/(1/3))/2$, DELTA= .2402 rROB.MECHNSM.≕ D1: A^2 FOR EA= 21 K/M FOR $A4:$ (-LN(1-A)) (1/4), DELTA=..5461 FOR A3: (-LN(I-A)}"(I/3), DELTA= .5052 FOR A2: $(-LN(1-A)) \cap (1/2)$, DELTA= .4235 FOR R2: $1-(1-A)/(1/2)$, DELTA= .2507 FOR R3: 1 (I-A) (i/3), DELTA= .2279 FOR FI: -LN(1-A). DELTA= .1797 FOR DI: A"2, DELTA= .0888 FOR D2: A+(I-A)LN(I-A>, DELTA= .1297 FOR D4= i-(2A/E)-(I-A) ~2/3)~ DELTA= .1546 cOR D3: (1-(1-A) ~(1/3)) ~2, DELTA= .2129 PRDB. MECHNSM. = D1: A"2 FOR EA= 22 + M FOR A4: (-LN(1-A)) (1/4), DELTA= .5734 FOR A3: $(-LN(1-A)) \wedge (1/Z)$, DELTA= .5325 FOR A2: (-LN(I-A))~(I/2)~ DELTA= .4508 FOR R2: 1-(1-A) ~(1/2), DELTA= .278 FOR R3: $1-(1-A)$ (1/3), DELTA= .2552 FOR FI: -LN(I-A), DELTA= .2068 FOR DI: A"2, DELTA= .0763 FOR D2: A+(1-A)LN(1-A), DELTA= .1046 FOR D4: $1-(2A/3)-(1-A)^{2}/(2/3)$, DELTA= .1279 FOR D3: (I-(I-A)~(I/3))'2, DELTA= .1856 F ROB.MECHNSM.= DI: A"2 FOR EA= 23 KIM FOR A4: (-LN(1-A))^(1/4), DELTA= .6007 FOR A3: (-LN(I-A))"(I/3), DELTA= .5598 FOR A2: (-LN(1-A)) ^(1/2), DELTA= .4781 FOR R2: I-(1-A)"(1/2), DELTA= .3052

FOR R3: **I-(l-A)"(l/') _ ,** DELTA= -2825 FOR F1: - LN(1-A), DELTA= .2339 FOR **Cl: A"2,** DELTA= .0723 FOR D2: A+(1-A)LN(1-A), DELTA= .0808 FOR D4: $1-(2A/3)-(1-A)^\wedge(2/3)$, DELTA= .1017 FOR D3: **(1-(1-A)' (l/3) j-2.** DELTA= .1584 PROB.MECHNSM.= Dl: A"2 FOR EA= 24 k/M FOR A4: (-LN(1-A)) (1/4), DELTA= .6279 FOR A3: $(-LN(1-A)) \cap (1/3)$, DELTA= .587 FOR A2: (-LN(l-A))' (l/2). DELTA= .5053 FOR R2: 1-(1-A:' (l/Z), DELTA= -3325 FOR R3: 1-(1-A)²(1/3), DELTA= .3098 FOR Fi: $-LN(1-A)$, DELTA= .261 FOR **Dl: A 2.** DELTA= -0783 FOR D2: A+(1-A)LN(1-A), DELTA= .0602 FOR D4: 1-(2A/3)-(1-A)^(2/3), DELTA= .0761 **FOR D:: ~:-(l-A)h(l/3))"2.** DELTA= **-1311 PRO~.MECHNSH.= E2:** A+(z-A)LN(i-A) FOR EA= **25 k/M** FOR A4: $(-LN(1-A)) \cap (1/4)$, DELTA= .6552 FOR A3: (-LN(1-A)) (1/3), DELTA= .6143 FOR A2: (-LN(1-A)) $(1/2)$, DELTA= .5325 FOR R2: **I-(I-A)"(?/2!.** DELTA= .3597 FCR R3: **1-Cl-AP(l/3),** DELTA= -337 COR Fl: -LN!l-A), DELTA= -2881 FOR **Dl: A.'2.** DELTA= -0923 FOR $D2$: A+(1-A)LN(1-A), DELTA= .0469 FOR D4: 1-(2A/3)-(1-A)^(2/3), DELTA= .0524 FOR C3: **Cl-Cl-A)"(1/31)-2,** DELTA= .1039 PROB.IlECHNSr!.= D2: **Attl-A)LN(l-A) FOR EA= 26 K/H** FOR A4: $(-LN(1-A))$ (1/4), DELTA= .6824 FOR A3: (-LN(1-A)) (1/3). DELTA= .5415 FOR A2: **(-LNCI-A))'(1/2!, DELTA= .5598 FOR R2: 1-(l-A)'(!/Z). DELTA= .3869** FOR R3: 1-(1-A)[~]'1/3). DELTA= .3643 FOR **Fl:** -LN(l-A). DELTA= -3153 \overline{P} OR D1: A $\overline{2}$, DELTA= .1113 FOR D2: A+(1-A)LN(1-A), DELTA= .0<mark>475</mark>
FOR D4: 1-(2A/3)-(1-A)^(2/3), DELTA= .034 SOR D3: $(1-(1-A)-(1/3))$, DELTA= .0766 $PROB. MECHNSM = D4: 1-(2A/3)-(1-A) (2/3) FOR E4 = 27 K/M$ **FOR** A4: **!-LNCI-A))- (:/4).** DELTA= .709e FOR A3: (−LN(1−A))^(1/3), DELTA= .66 F-OR A2: FOR R2: **(-LN(l-A)) "(l/2!, DELTA= .5f37 ?-(1-A) (l/2). DELTA= .4141** FOR R3: **1-Cl-A).'!l,'3) , DELTA= -3915** FOR Fl: **-LN (1 -A) . DELTA= -3424** FOR **Dl: A-2, DELTA= .I332** FOR D2: **A+!l-AlLN!l-A).** DELTA= .0617 FOR D4: 1-(2A/3)-(1-A) '(2/3), DELTA= -033 FDR D3: **(1-(1-A) '(l/3))'2, DELTA= .0494** PROB.MECHNSM.= D4: $1-(2A/3)-(1-A)+(2/3)$ FOR EA= 28 K/M FOR A4: **!-LN(l-A))' !1/4), DELTA= .7369 FOR A;:** (-LN(i-A) **)-'(l/3) *** DELTA= -596 FOR AZ: **:-LE?(l-A) > ((l/Z). DELTA= .&I42** FOR R2: **1-(1-A) '(l/2).** DELTA= -4414 FOR R3: 1-(1-A) '(l/3), DELTA= .4187 FOR **Fl: -LN (1-A) .** DELTA= _ 3696

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FOR D1: A 2, DELTA= . 1568
FOR D2: A+(1-A)LN(1-A), DELTA= .0826
FOR D4: 1-(2A/3)-(1-A) (2/3), DELTA= .0499
TOR D3: (1-(1-A) (1/3)) 2. DELTA= .0222
PROB.MECHNSM.= D3: (1-+1-A) (1/3)) 2 FOR EA= 29 K/M
FOR A4: (-LN(1-A)) (1/4), DELTA= .7641
FOR A3: (-LN(1-A)) (1/3), DELTA= .7232
FOR A2: (-LN(1-A)) (1/2), DELTA= .6414
FOR R2: 1-(1-A) (1/2), DÉLTA= .4686<br>FOR R3: 1-(1-A) (1/3), DELTA= .4459
FOR Fi: - LN(1-A), DELTA= .3967
FOR Di: A 2, DELTA= .1814
FOR D2: A+(1-A)LN(1-A), DELTA= .1065<br>FOR D4: 1-(2A/3)-(1-A) (2/3), DELTA= .0733
FOR DE: (1-(1-A)-(1/3)) 2, DELTA= 5.1E-03
PROB.MECHNSM. = D3: (1-(1-A) (1/3)) 2 FOR EA= 30 F/M
FOR A4: (-LN(1-A)) (1/4), DELTA= . 7913
FOR AC: (-LN(1-A)) (1/C), DELTA= .7504
FOR A2: (-LN(1-A)) (1/2), DELTA= .6687
FOR R2: 1-(1-A) (1/2), DELTA= .4959
FOR R3: 1-(1-A) (1/3), DELTA= .4731
FOR F1: - LN(1-A), DELTA= . 4239
FOR D1: A 2, DELTA= .2066
FOR D2: A+ 1-A)LN(1-A), DELTA= .1316
"OR D4: 1-(2A/3)-(1-A) (2/3), DELTA= .0986
FOR DD: (1 - (1 - A) - (1/3)) 2, DELTA= .0323
PROB.MECHNSM.= D3: (1-(1-A) - (1/3)) 2 FOR EA= 31 F/M
EA= 50 FEAL/MOL FOR MECHNOM, D3: (1-x1-A) (1/3)) 2, DELTA= 5.06464008E-03
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